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**REASONING TESTS**

**Questions, Tips and  
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long is a C-C single  
bond of an alkane?

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None of these.

Answer: A Learning

Objective: 2.2

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Difficulty: Easy. 35.

Which of the following is the correct bond-line structure for (CH.

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Questions The

Covalent Bond 1. The

hybridization of the

central carbon in

CH<sub>3</sub>C≡N and the

bond angle CCN are

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a.  $sp^2$ ,  $180^\circ$ . b.  $sp$ ,  $180^\circ$ . c.  $sp^2$ ,  $120^\circ$ . d.  $sp^3$ ,  $109^\circ$ . 2. Which of the following statements about an  $sp$  hybridized carbon is FALSE? a. It is divalent. b. It forms bonds that are linear. c. It has two p orbitals. d.

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isomers; different  
compounds entirely  
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