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Second Edition. H. Scott Fogler. Ame and Catherine Vennema Professor of Chemical Engineering and The Arthur F. Thurnau Professor at the University of Michigan, Ann Arbor, Michigan.

Essentials of Chemical Reaction
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Solution The general solution method is best explained in terms of the complex reaction network notation where the rate for the j th species is given : $N_k \frac{dy_j}{dt} = \sum_{i=1}^I \nu_{ij} r_i$ (1) A solution is assumed of the form: $y_j = Y_j e^{(2)}$ where Y_j and are constants to be determined.

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The PFR volume required in reaching $X=0.5$ (reaction rate is independent of conversion). The PFR volume required to go from $X=0.5$ to $X=0.7$ (reaction rate depends on conversion). Finally, we add V_2 to V_1 and get: $V_{tot} = V_1 + V_2 = 2.3 \times 10^{11} \text{ m}^3$.

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