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The Rosen theory was used to describe the synergism in the surfactant mixtures. The surfactant compositions reveal a high usefulness to cleaning oil contaminations. In order to obtain a recipe for an effective degreasing agent, a selection of suitable surfactants is required.

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Milton J. Rosen, PhD, is Professor Emeritus of Chemistry at Brooklyn College of the City University of New York. He is also the Director (ret.) of the university's Surfactant Research Institute, a pioneering

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organization that he founded in 1987.

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1 Surfactants and Their Solutions: Basic Principles Laurier L. Schramm^{1,2} and D. Gerrard Marangoni³ ¹Petroleum Recovery Institute, 100, 3512 - 33rd St. NW, Calgary, AB, Canada T2L 2A6 ²University of Calgary, Dept. of Chemistry, 2500 University Drive NW, Calgary, AB, Canada T2N 1N4 ³St. Francis Xavier University, Dept. of Chemistry, PO Box 5000, Antigonish, NS, Canada B2G 2W5

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Plots of the surface or interfacial tension of aqueous surfactant

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solutions versus the logarithm of the bulk concentration of the surfactant frequently become linear at concentrations from 10 to 30% of the critical micelle concentration. According to the Gibbs equation, the surface excess concentration of the surfactant in the linear region is constant; saturation adsorption is said to exist.

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